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SPECIFICATION

TO ALL WHOM IT MAY CONCERN:

Be it known that I,

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have invented certain improvements in

“DIESEL ENGINE FUEL IN MICROEMULSION FORM AND METHOD FOR PREPARING IT”

of which the following description is a specification.

[illegible]

BACKGROUND OF THE INVENTION

The present invention relates to a fuel in microemulsion form, particularly for supplying Diesel engines, and to a method for preparing it.

5 Emulsions or microemulsions of petroleum products and water in which particular surfactants or mixtures of surfactants are used are known in the art.

For example, US-3,876,391 discloses microemulsions of hydrocarbon products with water in which a mixture of surfactants is used which is
10 constituted by a first surfactant which is soluble in the oil phase and a second surfactant which is soluble in the aqueous phase, to which a further water-soluble additive, for example an amide, an alkanolamine, a polyamine or an aldehyde, is added.

US-4,465,494 discloses microemulsions of liquid fuels and water which
15 contain an alcohol or an amine and, as surfactant, a salt of an alkylphenoxyalkanoic acid.

A fuel emulsified with water is disclosed in EP-630,398 and is obtained by mixing the compounds in a static mixer in particular pressure and temperature conditions in the presence of a mixture of surfactants
20 constituted by sorbitan oleate, a polyalkylene glycol and an alkylphenol ethoxylate.

In general, the use of surfactants or other additives such as the ones mentioned above can entail problems both because they can be inherently corrosive with respect to the devices with which they come into contact and
25 because toxic by-products can form during combustion.

Moreover, on the basis of the experience of the Applicant, emulsions of liquid fuels and water prepared according to conventional methods by adding appropriate surfactants generally entail stability problems even in optimum storage conditions, so that after a certain time an at least partial
30 separation of the phases is observed which entails many drawbacks during

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the combustion process owing to the non-uniformity of the supplied fuel.

EP-372,353 by the same Applicant discloses a stabilized emulsion of a fuel, particularly a fuel for Diesel engines, and water, with the addition of a product which acts as a lubricant and antifreeze, for example sorbitol monoleate. The stabilized emulsions of fuel are prepared by using a turbine-effect emulsifier such as the one disclosed in EP-124,061 in the name of this same Applicant.

Applicant has noted that in some cases, particularly in case of use of low-density fuels, the preparation of the fuels as disclosed in EP-124,061 and EP-372,353 entails a relatively high energy expenditure and a reduction in the productivity of the system, if one seeks high productivity and stability.

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A method with improved efficiency for forming the emulsion of a fuel is disclosed in co-pending EPA No. 00121331.3 and corresponding co-pending US Application by the same Applicant, included herein by reference. This method requires the use of an apparatus having a particular geometry.

The above applications disclose an apparatus and a method for forming stabilized atomized microemulsions from different liquids which are normally immiscible; the apparatus comprises a primary chamber and a sequence of at least two cavitation chambers arranged in succession, means for feeding primary and secondary fluids into the primary chamber, and means for the exit of the formed microemulsion from the last cavitation chamber, the primary chamber and the cavitation chambers being fluid-connected to each other by way of fluid passage means which are adapted to produce a velocity of the fluids, during passage through the passage means, which gradually increases from the primary chamber toward the last cavitation chamber. The method according to the above applications comprises the stage of premixing the primary fluid with the secondary fluid, followed by the passage of the premix of fluids through a succession of steps of flow at a higher velocity alternated with steps of flow at a lower

velocity, the higher flow velocities gradually increasing from the first higher-velocity step to the last higher-velocity step.

SUMMARY OF THE INVENTION

The aim of the present invention is to provide a fuel, particularly for Diesel engines, in microemulsion form which produces, in exhaust gases, a 97% reduction in the grade of smoke, a reduction of the carbon oxide content of more than 50%, a reduction of more than 1% in carbon dioxide, and a reduction in nitrogen oxides of more than 35-40% with respect to the conventional base fuel.

An object of the present invention is to provide a fuel in microemulsion form which is simple to prepare, for example by mixing the components with minimal agitation, without any need to follow a preferential order in adding the components.

Another object of the present invention is to provide a fuel in microemulsion form which poses no engine operation problems and can be stored in storage tanks in the same conditions as Diesel fuel.

Another object of the present invention is to provide a fuel for use in feeding Diesel engines, such as engines for medium-heavy transport, engines for heavy transport, marine engines, electric power generators and turbines and can also be used in civil heating applied to gas oil burners.

DETAILED DESCRIPTION OF THE INVENTION

This aim and these objects and other aims and objects which will become better apparent hereinafter from the following description are achieved by a fuel, particularly for Diesel engines, in microemulsion form, which comprises a liquid fuel, an emulsifier or primary activating agent, an emulsive agent or secondary activating agent and water, said emulsive agent having a suitable HLB (Hydrophilic-Lipophilic Balance) value which is preferably higher than 9.

Preferably, the liquid fuel is a liquid fuel for Diesel engines, particularly Diesel fuel.

The Diesel fuel used to obtain the fuel according to the present invention can be Diesel fuel for automotive applications but also a Diesel fuel for different uses, including arctic Diesel fuel and winter Diesel fuel.

The emulsifier used to form the fuel according to the present invention
5 can be a sorbitan monoleate and is preferably a sorbitan monoleate having
the characteristics given in Table 1.

Table 1: Sorbitan monoleate characteristics

App�arance at 20�C	Oily
Saponification number (mg KOH/g)	145-165
Color	light amber (Gardner 10 max)
Acidity index	7 max
Odor	sweet, fatty
Hydroxyl index (mg KOH/g)	190-215
Evolution temperature/range	Decomposes
Melting point	-13�C pouring temperature
Flammability point	> 200�C Pensky Martens method, closed cup
Self-ignition temperature	> 200�C
Non-explosive	
Density	1.010-1.040 g/cm ³ at 25�C
Can be dispersed in water, non-soluble	
pH 1% in water	neutral (approximately 7)
Viscosity	970-1080 mPa.s at 25�C
Acute toxicity	> 5000 mg/kg (rat)

10 As an alternative, in order to obtain the fuel according to the present invention it is possible to use other emulsifiers which are in any case still definable as sorbitan monoleate also of the hydroxystearate type, even with

a saponification number and a hydroxyl number which are lower than the ones indicated in Table 1 and an acidity number which is higher than the one indicated in Table 1.

The inventor of the present invention has found that the presence of the emulsive agent is fundamental both when the preferred emulsifier as indicated in Table 1 is used and when another alternative emulsifier is used.

Preferably, the emulsive agent is constituted by nonylphenol ethoxylate, which can also be defined as polyethylene glycol isononyl phenyl ether or as isononyl phenol ethoxylate, nonylphenol polyglycol ether, alkylphenol polyglycol ether, even more preferably with the characteristics stated in Table 2.

Table 2: characteristics of emulsive agent

No. of moles of ethylene oxide	6
Physical appearance at 25°C	clear colorless liquid
State transition	< approximately -10°C - MPL 1001,1
Turbidity point (10% in BDG at 25%)	68 to 69°C MPL 2001,0
Hydroxyl index	115 to 121 mg/KOH/g - MPL 1010,0
Average relative molecular mass	464 to 487 - calculated value
Free polyethylene glycols	>= approximately 3% by weight - MPL 2002,0
pH (5%)	5 to 7 - MPL 1007,0
Water (Karl-Fisher)	> = approximately 0.5% by weight, calculated
Ash	> = approximately 0.2% by weight, calculated
HLB	10.7, calculated

The inventor of the present invention has found that a fundamental characteristic of the emulsive agent comprised in the fuel according to the present invention is the emulsive capacity in order to allow the integration of water at a high level.

5 The emulsive capacity required for the purposes of the present invention is achieved by emulsive agents with a calculated HLB higher than 9, preferably higher than 10,7 for better hydrophilic properties. HLB (hydrophilic-lipophilic balance) reflects the balance of the hydrophilic-lipophilic properties of the emulsive agent and is determined with
10 conventional methods which are typical in the chemical production field.

The inventor of the present invention has found that such an HLB value is reached by a nonylphenol ethoxylate having the properties indicated in Table 2, and in particular with a number of ethylene oxide moles of no less than 6 (number of moles, i.e., amount of ethylene oxide material introduced
15 in the manufacturing process).

The nonylphenol ethoxylate used to obtain the fuel according to the present invention preferably has the chemical formula $R-C_6H_4-(O-CH_2-CH_2)_nOH$ -->, where $R = C_9H_{19}$ and n is approximately 6.

Even more preferred emulsive agents, also with reference to possible
20 future statutory aspects, are non-ionic surfactants such as for example C_{12} C_{13} alcohol ethoxylate with an average of 8 moles of ethylene oxide, in particular having the following characteristics:

Physical appearance at 25°C	almost colorless turbid liquid
State transition at °C	15-20
Turbidity point °C	57-59
Number of hydroxyls mg KOH/gr	99-107
Average relative molecular mass	524-567
Free polyethylene glycols (% by weight)	≤ 3
pH (5%)	5-7

Water (Karl Fisher method) %	≤ 0.5
Ash % by weight	≤ 0.2
HLB	12.8

Moreover, a preferable emulsive agent is a C_{16} C_{18} cetyl stearyl alcohol ethoxylate with 11 moles of ethylene oxide of the non-ionic type, which can be combined with anionic and cationic surfactants, for example having the following characteristics:

Cetyl stearyl alcohol (fatty alcohol)	C_{16} C_{18} -11 moles of ethylene oxide
pH sol. 3%	5-7
NaCl 10% turbidity point	58-62°C
Hydroxyl number	69-75 mg KOH/g
Water content	$\leq 1\%$
Melting point	37.5-39.5°C
Acidity number	≤ 1
Relative density at 70°C	0.962-0.965 g/cm ³
Solidification point	$\leq 35^\circ\text{C}$
Flash point	$\leq 250^\circ\text{C}$
Solubility	clear solution in water at 40°C partial solution in water at 20°C
Appearance	solid
Color	whitish
Odor	almost non-existent (odorless)

Also these "emulsive agents" can be introduced in the same ratios as those provided for nonylphenol ethoxylate, i.e., from 5 to 20 parts in 1000 parts of intact Diesel fuel.

The presence of the emulsive agent C_{16} C_{18} -11 moles of ethylene oxide, when the microemulsion is processed in particular through the EMDT5 system provided with multiple reverse-flow coaxial turbines according to EPA No. 00121331.3 and to the corresponding co-pending US Application

by the same Applicant, allows to form stable microemulsions, such as environment-friendly fuels for Diesel engines, even with demineralized water up to 15% by weight of bio-vegetable fluids, such as for example biodiesel or methyl esters composed of rapeseed and sunflower oil which
5 are commercially available and cannot be used in their natural state as fuels for Diesel engines since they cause seizure and produce bad odors.

The presence of emulsive agents such as $C_{12}C_{13}$ and $C_{16}C_{18}$ alcohols with 8 and 11 moles of ethylene oxide is therefore a significant help also in significantly reducing the "French-fries" odor that is typical of the
10 combustion of a methyl ester in its natural state, in addition to more easily assisting the presence of an antifreeze such as monoethylene glycol, even at 3 to 6% with respect to the percentage of water; the presence of a biocide for "antibacterial protection" in order to provide "resistance to bacterial and fungal attack" is also helpful.

15 Preferably, the fuel in microemulsion form according to the present invention comprises, in parts per volume, for 1000 parts of conventional Diesel fuel, 13 to 17 parts of emulsifier, 5 to 20 parts of emulsive agent, and 100 to 145 parts of water.

The water used is preferably demineralized water, but it is also possible
20 to use water from the water mains, preferably filtered and at a maximum temperature of 40-45°C.

The fuel according to the present invention can contain, preferably in an amount between 400 and 800 ppm, products based on polyisobutylene succinic anhydride and also specific biocide products.

25 The addition of products based on polyisobutylene succinic anhydride meets a possible requirement of improving the cetane number (owing to the presence of water) and also of giving better characteristics toward low temperatures. The biocides meet the optional need to avoid the formation of bacterial colonies due to the presence of the sorbitan monoleate and due to
30 the natural bacterial presence in Diesel fuel containment tanks, which grows

especially in hot climates ($>27^{\circ}\text{C}$).

The fuel according to the present invention can be obtained by mixing the components, i.e., conventional base fuel, preferably Diesel fuel, emulsifier, emulsive agent and water, with minimal agitation. The formation
5 of the microemulsion is practically instantaneous and is revealed by an instantaneous change of color of the mixture of components, which becomes white. The instantaneous behavior is also an essential visual parameter for determining the result.

The microemulsion is formed even as easily as by placing the
10 components in a container, even a bucket, and by performing minimal agitation. The microemulsion is characterized by microcells having dimensions substantially smaller than $0.15\ \mu\text{m}$.

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C7/ The resulting microemulsion is stable even after centrifugation at over
35,000 m/s^2 .

15 The preparation of the fuel according to the invention, which is termed GECAME-2, occurs substantially spontaneously or in any case with minimal agitation of the components, without the need for devices such as the turbine-effect emulsifier. However, for the sake of high productivity in a particular short time, the fuel according to the invention can be formed by
20 using an apparatus such as the one disclosed in co-pending EPA No. 00121331.3 and corresponding co-pending US Application by the same Applicant. The use of such a device is suggested merely owing to the fact that this system allow homogenization of large volumes substantially
instantaneously with very low industrial costs.

25 The disclosures in Italian Patent Application No. MI99A002393 from which this application claims priority are incorporated herein by reference.